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H<sub>3</sub>PO<sub>2</sub> (acids of phosphorous)

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Differences between  
acid and bases. Acids

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Acids are substances that produce hydrogen ions in a solution, while bases are considered substances capable of dissociating hydroxide ions in a solution.

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What are the differences between acids and bases? |

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taking place during  
the experiment.

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Mineral acids are  
acids prepared from

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minerals. For

example,

Hydrochloric acid

(HCl), Sulphuric Acid

(H<sub>2</sub>SO<sub>4</sub>), and nitric

acid (HNO<sub>3</sub>) etc. Also

Check Dilute Acids.

Bases. The most

common

characteristic of

bases is their bitter

taste and soapy feel.

A base is a substance

that renders hydroxyl

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ion(OH<sup>-</sup>) in their  
aqueous ...

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Acids and bases are  
used around the  
house for cleaning,  
but the levels of acids  
and bases in the body  
are also important 1.

The body needs a

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neutral pH and controls the amounts of acids and bases through processes involving the kidneys, lungs and blood 1. Significance. Acids and bases function to balance the pH levels in the body 1.

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Importance of Acids

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An acid is a  
compound that  
yields hydrogen ion  
( $H^+$ ) in aqueous  
solution and a base is  
a compound that  
gives hydroxyl ( $OH^-$ )

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ions in aqueous solution. As defined by Arrhenius, acid-base reactions are characterized by acids, which dissociate in aqueous solution to form hydrogen ions ( $H^+$ ) and bases, which form hydroxide ( $OH^-$ ) ions.

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