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CI 12 Math Ncert Solution

A solution of $[\text{Ni}(\text{H}_2\text{O})_6]^{2+}$ is green but a solution of $[\text{Ni}(\text{CN})_4]^{2-}$ is colourless. Explain. Explain. Ans: In $[\text{Ni}(\text{H}_2\text{O})_6]^{2+}$, Ni is in + 2 oxidation state and having 3d 8 electronic configuration, in which there are two unpaired electrons which do not pair in the presence of the weak H_2O ligand.

NCERT Solutions For Class 12 Chemistry Chapter 9 ...

NCERT TEXTBOOK QUESTIONS SOLVED. 2.1. Calculate the mass percentage of benzene (C₆H₆) and carbon tetrachloride (CCl₄) if 22 g of benzene is dissolved in 122 g of carbon tetrachloride.

Ans: Mass of solution = Mass of C₆H₆ + Mass of CCl₄ = 22 g + 122 g = 144 g
Mass % of benzene = $\frac{22}{144} \times 100 = 15.28\%$

NCERT Solutions For Class 12 Chemistry Chapter 2 Solutions

The amount of ion concentration in the solution is the ionic strength of the solution. It is articulated as I. The ion activity is affected by it. It is denoted with the ion interaction with water and other ions in the solution.

Ionic Strength Formula with Solved Questions

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 $2\text{Na} + 2\text{CH}_3\text{CH}_2\text{CH}_2\text{Cl} \rightarrow \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3 + 2\text{NaCl}$... Test paper with Video Solution Register and Get connected with IITian Chemistry faculty ...

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Potassium is an alkali metal and is a part of group 1. Its symbol is K taken from its Latin name Kalium. Its atomic number is 19 and atomic weight is 39.098u. It is white with a silvery shine or luster. It is soft at room temperature. Potassium is a fine conductor of electricity and heat. Visit Vedantu.com to read more about the Physical and Chemical Properties of Potassium.

Potassium | Physical and Chemical Properties of Potassium

Cl = 2, 8, 7 So it contains 7 valence electron. In order to achieve the nearest noble gas configuration, it gains one electron to form Chloride ion. Cl⁻ = 2, 8 An Ionic bond is formed between sodium ion and chloride ion by complete transfer of electron from sodium to chlorine. b. Atomic number of Magnesium(12) atom is 12. Electronic ...

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Revision Notes on Coordination Compounds. Ligands: an ion or molecule capable of donating a pair of electrons to the central atom via a donor atom. Unidentate ligands: Ligands with only one donor atom, e.g. NH₃, Cl⁻, F⁻ etc. Bidentate ligands: Ligands with two donor atoms, e.g. ethylenediamine, C₂O₄²⁻ (oxalate ion) etc. Tridentate ligands: Ligands which have three donor atoms per ligand, e ...

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A 5 % solution of sodium chloride means that 5 g of NaCl is present in 100g of the solution. 2. Volume

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percent is the number of units of volume of the solute per 100 units of the volume of solution. A 5 % (v/v) solution of ethyl alcohol contains 5 cm³ of alcohol in 100 cm³ of the solution.

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