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of mass. states that mass is neither created nor destroyed during ordinary reactions or physical changes.

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

~~3 Atoms: The Building Blocks of Matter~~

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Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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is neither created nor destroyed during ordinary chemical reactions or physical changes $H_2 + O = H_2O$. law of definite proportions. Holt Mcdougal Modern Chemistry

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

~~3 Atoms: The Building Blocks of Matter~~

Modern Chemistry 105 Chapter Test Name _____ Class _____ Date _____ Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from $10^\circ C$ to $90^\circ C$. According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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Standardized Test Prep CHAPTER REVIEW To give students practice under more realistic testing conditions, give them 60 minutes to answer all of the ques- tions in this Standardized Test Preparatio Answer the following items on a separate piece of paper. MULTIPLE CHOICE I. In the modern periodic table, elements are arranged according to

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